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Chemistry If8766 Instructional Fair Inc - WordPress.com Molarity is expressed in units of moles per liter (mol/L). It's such a common unit, it has its own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

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where the subscripts "1" and "2" refer to the solution before and after the dilution, respectively. Since the dilution process does not change the amount of solute in the solution, $n_1 = n_2$. Thus, these two equations may be set equal to one another:

$[M_1L_1=M_2L_2]$ This relation is commonly referred to as the dilution equation.

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Molarity is expressed in units of moles per liter (mol/L). It's such a common unit, it has its own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

Molarity Definition as Used in Chemistry

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